Seat Number PANKH-26 BP-203T Total Pages : 5] Biochemistry Time: 3 Hours (712203)Note : (1) Do not write anything on question paper except seat no. Max Marks: 75 (2)Graph or diagram should be drawn with the black ink pen being used for writing paper or black HB pencil. (3)Students should note, no supplement will be provided. (4)Figures to the right indicate full marks. All questions are compulsory. (5)Multiple Choice Questions: 1. 20 (i)The most important sugar concerned with human biochemistry is: (a) Cane sugar **(b)** Glucose (c) Lactose (d)Fructose (ii)Which one of the following is not a disaccharide? (a)Sucrose **(b)** Lactose (c) Maltose (d) Starch The net number of moles of APP produced when one free glucose molecule (iii) undergoes glycosis under aerobic condition is : 2 (a) (b) 3 8 (d)9 (c)

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(iv)	En	Strone involving fran	efer of elec-	trons	of the hydrogen atoms of the	
		strate to oxygen are				
	(a)	Oxygenases		(b)	Oxidases	
	(c)	Dehydrogenases		(d)	Hydroperoxidases	
(v)	The	The common currency of energy in biological reaction is :				
	(a)	ATP		(b)	AMP	
	(c)	ADP		(d)	UDPG	
(vi)	Whic	h of the statements	regarding	coe	nzyme is incorrect ?	
	(a)	Every coenzyme is	a cofactor	and	every cofactor is a coenzyme.	
	(b)	Every coenzyme is a	cofactor b	ut ev	very cofactor is not a coenzyme.	
	(c)				utents of enzymes.	
	(d)				eotides and are composed of	
		vitamins.	,			
(vii) (Oxidas	ses are the enzyme	s which :			
				ogen	from one substrate and pass	
,,		t second substrate.		ogen	nom one case	
(b					form a substrate and pass it	
(0			al of hydro	ogen	from a substrate and pass it	
24.		irectly to oxygen.				
(c)	C	atalyse the incorpo	ration of	oxyg	en directly into the substrate.	
(d)) A	ll the three.				
(viii) Th	e nur	nber of different a	mino acid	ls fo	und to be present in natural	
		are:				
(a)	10	i	(b	.)	15	
					10	
(c)	25		(a	<i>l</i>)	20	
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	(ix)	Amir	no ncids exists ns :						
		(a)	Cations						
		(c)	Zwitter ions	(b)	Anions				
	(x)	The		(d)	None of these				
		The metabolism of protein is integrated with that of carbohydrate and fat through:							
		(a)	Malate						
		(c)	Isocitrate	(b)	Citrate				
	(xi)	Whic		<i>(d)</i>	Oxaloacetate				
	(a)	ch of the following cannot Threonine	transamination ?						
			- meoning	(b)	Alanine				
		(c)	Serine	(d)	Valine				
	(xii)	Lipids are:							
		(a)	Soluble in water	(b)	Soluble in organic solvents				
		(c)	Soluble in both	(d)	Insoluble in both				
	(xiii)	Fatt	Fatty acids are oxidised mainly by:						
		(a)	β -oxidation	(b)	α-oxidation				
		(c)	γ -oxidation	(d)	All of these				
	(xiv)	When the concentration of ketone bodies in human blood increases, the							
			lition is known as :						
		(a)	Ketonuria	(b)	Ketonemia				
		(c)	Ketosis	(d)	Ketogenesis				
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(tv)	Disorders of lipid metabolism includes:						
	(a)	Fatty liver	(b)	Obesity			
(H-26 (m)	(c)	Atherosclerosis	(d)	All of these			
(xvi	The three common bases in DNA and RNA are :						
	(a)	Adenine, Guanine	and Cytosine				
	(b)	Adenine, Guanine	and Uracil				
	(c)	Adenine, Guanine	and Thymine				
	(d)	None of the above					
(wii)) The	disease gout is char	acterised by in	ncreased level of :			
	(a)	Uric acid	(b)	Allantoin			
	(c)	Ammonia	(d)	Creatinine			
Caviii) Ene	rgy is measured in w	hich of the fo	ollowing units ?			
	(a)	Kelvin	(b)	Joule			
	(c)	Pascal	(d)	Mol			
(xix)	Enthalpy is represented by which of the following symbols?						
	(a)	Н	(b)	K			
	(c)	S		U			
(xx)	The	type of Sugar is nuc					
	(a)		rete delus are	:			
		Hexose	(b)	Tetrose			
	(c)	Heptose	(d)	Pentose			
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2.	Solve	any	two	of	the	following	:
2.	Solve	any	two	ot	the	following	

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- (a) Explain Watson and Crick DNA model and add a note on DNA replication.
- (b) Explain in detail about transamination and reactions of Urea cycle.
- (c) Explain the various disorders related to lipid metabolism.
- 3. Write short notes on the following (any seven):

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- (a) Electron transport chain.
- (b) Energy rich compounds.
- (c) Classification of enzymes according to IUB.
- (d) Ketone bodies.
- (e) Catabolism of Phenylalanine and Phenyl ketonuria.
- (f) Hyperuricemia and Gout.
- (g) Enzyme inhibitors.
- (h) Enthalpy and Entropy.
- (i) Endergonic and Exergonic reaction.